

Turbojet Aircraft Engine

An aircraft operates on this thermodynamic cycle.



Get properties from ThermophysicalData package

Property := ThermophysicalData:-Property

Atmosphere := ThermophysicalData:-Atmosphere

 ${\sf TemperatureEntropyChart} \coloneqq {\sf ThermophysicalData:-TemperatureEntropyChart}$

Parameters

Velocity of aircraft and altitude

$$v_{aircraft} := 200 \text{ m} \cdot \text{s}^{-1}$$
 $h_{aircraft} := 12 \text{ km}$

Heating value of fuel

$$H_{fuel} := 4.38 \times 10^7 \, J \cdot kg^{-1}$$

Temperature at point 3

Compression ratio

 $CPR \coloneqq 6.5$

Efficiencies

$\eta_{\text{combustion}} := 0.99$	$\eta_{turbine} := 0.9$	$\eta_{\text{jetpipe}} \coloneqq 0.95$
$\eta_{\text{compressor}} \coloneqq 0.855$	$\eta_{\text{drive}} \coloneqq 0.98$	$\eta_{\text{intake}} \coloneqq 0.9$

Area of nozzle

$$A_{nozzle} \coloneqq 0.168 \, m^2$$

Pressure loss in combustion chamber

$$\Delta \mathsf{P}_{\mathsf{chamber}} \coloneqq \mathsf{0.05793}\,\mathsf{bar}$$

Gravity

$$\mathbf{g} \coloneqq 9.81 \, \mathbf{m} \cdot \mathbf{s}^{-2}$$

Gas constant

$$R_0 := Property(gasconstant, air, useunits) = 8.315 \frac{J}{mol \cdot K}$$

Molar mass of dry air

$$M_{DryAir} := Property(MOLARMASS, air, useunits) = 2.90 \times 10^{-2} \frac{kg}{mol}$$

Air Intake to Compressor Inlet

Air temperature and pressure

$$T_0 := Atmosphere(h_{aircraft}, temperature) = 216.650 K$$

$$p_a := Atmosphere(h_{aircraft}, pressure) = 1.933 \times 10^4 Pa$$

Kinetic energy of air

KINELIC ENERGY OF an

$$KE := 0.5 \cdot v_{aircraft}^{2} = 20.00 \frac{kJ}{kg}$$

Entropy and enthalpy of air

$$s_{0} := \operatorname{Property}(\operatorname{entropy}, \operatorname{air}, \operatorname{temperature} = T_{0}, \operatorname{pressure} = p_{a}) = 4.036 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}$$
$$h_{0} := \operatorname{Property}(\operatorname{enthalpy}, \operatorname{air}, \operatorname{temperature} = T_{0}, \operatorname{pressure} = p_{a}) = 342.813 \frac{\text{kJ}}{\text{kg}}$$

. .

Assuming air intake is isentropic, entropy at compressor inlet

$$\mathbf{s}_1 := \mathbf{s}_0$$

Assuming all the kinetic energy turns into sensible heat, the enthalpy at compressor inlet

$$h_1 := h_0 + KE = 3.628 \times 10^5 \frac{m^2}{s^2}$$

Temperature at compressor inlet

$$T_{1'} := Property(temperature, air, enthalpy = h_1, entropy = s_1)$$

$$T_1 := \eta_{intake} \cdot (T_{1'} - T_0) + T_0 = 234.614 \text{ K}$$

Pressure at compressor inlet

$$p_1 := Property(pressure, air, temperature = T_1, entropy = s_1) = 25.529 kPa$$

Compressor Inlet to Compressor Outlet

Entropy at compressor outlet

 $\mathbf{s_2} := \mathbf{s_1}$

Pressure and temperature at compressor outlet

$$\mathsf{p}_2 := \mathsf{CPR} \cdot \mathsf{p}_1 = \ 1.659 \times 10^5 \, \mathsf{Pa}$$

 $T_2 := Property(temperature, air, pressure = p_2, entropy = s_2) = 400.313 K$

$$T_2 - T_1$$

$$T_{2'} := \frac{2}{\eta_{compressor}} + T_1 = 428.414 \text{ K}$$

Enthalpy at compressor outlet

$$h_2 := Property(enthalpy, air, temperature = T_2, pressure = p_2) = 556.061 \frac{kJ}{kg}$$

Combustor Inlet to Outlet

Pressure loss over combustor

$$\mathbf{p}_3 := \mathbf{p}_2 - \Delta \mathbf{P}_{chamber} = 1.601 \times 10^5 \, \mathbf{Pa}$$

Entropy and enthalpy at outlet

$$s_3 := \text{Property}(\text{entropy, air, } T = T_3, P = p_3) = 5.181 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}$$

 $h_3 := Property(enthalpy, air, temperature = T_3, pressure = p_3) = 1.343 \times 10^3 \frac{kJ}{kg}$

Turbine Inlet to Outlet

For an isentropic turbine

$$s_4 := s_3$$

Temperature at outlet

$$T_{4'} := Property\left(temperature, air, enthalpy = h_3 - \frac{h_2 - h_1}{\eta_{drive}}, entropy = s_4\right) = 976.508 \text{ K}$$

Using isentropic efficiency for turbine, pressure and temperature at outlet

$$T_4 := T_3 - \frac{T_3 - T_{4'}}{\eta_{turbine}} = 957.494 \text{ K}$$

 $p_4 := Property(pressure, air, temperature = T_4, entropy = s_4) = 77.527 kPa$

Nozzle

For choked flow in the nozzle (assuming isentropic operation)

$$\gamma_{\text{expand}} := \frac{\text{Property}(C, \text{air, pressure} = p_{4'} \text{ temperature} = T_{4'} \text{ useunits})}{\text{Property}(Cvmass, \text{air, pressure} = p_{4'} \text{ temperature} = T_{4'} \text{ useunits})} = 1.340$$

$$p_{c} := p_{4} \cdot \left(\frac{2}{\gamma_{expand} + 1}\right)^{\frac{\gamma_{expand}}{\gamma_{expand} - 1}} = 41.763 \text{ kPa}$$

 $ifelse(p_a < p_{c'} "sonic", "subsonic") = "sonic"$

$$T_{5'} := rac{2}{\gamma_{expand} + 1} \cdot T_{4'}$$
= 834.791 K

$$\mathsf{T}_{\mathsf{5}} := \mathsf{T}_{\mathsf{4}'} - rac{\mathsf{T}_{\mathsf{4}'} - \mathsf{T}_{\mathsf{5}'}}{\eta_{\mathsf{jetpipe}}} = 827.332 \ \mathsf{K}$$

Jet Pipe

$$p_{5} := p_{4} \cdot \left(\frac{T_{5}}{T_{4'}}\right)^{\frac{\gamma_{expand}}{\gamma_{expand} - 1}} = 40.310 \text{ kPa}$$

$$R := \frac{R_0}{M_{DryAir}} = 287.049 \frac{J}{kg \cdot K}$$

Specific volume

$$v_{5'} := \frac{R \cdot T_{5'}}{p_5} = 5.945 \frac{m^3}{kg}$$

Jet gas velocity

$$C_{jet} := \sqrt{\gamma_{expand} \cdot R \cdot T_{5'}} = 566.556 \frac{m}{s}$$

Mass flow

$$\mathsf{Mflow} := \frac{\mathsf{A}_{\mathsf{nozzle}} \cdot \mathsf{C}_{\mathsf{jet}}}{\mathsf{V}_{\mathsf{5}'}} = 16.011 \ \frac{\mathsf{kg}}{\mathsf{s}}$$

Total thrust

$$\mathsf{Thrust}_{\mathsf{momentum}} := \mathsf{Mflow} \cdot \left(\mathsf{C}_{\mathsf{jet}} - \mathsf{v}_{\mathsf{aircraft}}\right) = 5.869 \, \mathsf{kN}$$

$$\text{Thrust}_{\text{pressure}} \coloneqq (p_5 - p_a) \cdot A_{\text{nozzle}} = 3.525 \times 10^3 \, \text{N}$$

$$\mathsf{Thrust}_{\mathsf{total}} \coloneqq \mathsf{Thrust}_{\mathsf{momentum}} + \mathsf{Thrust}_{\mathsf{pressure}} = 9.394 \, \mathsf{kN}$$

Heat supplied

$$\mathbf{Q}_{\mathsf{input}} := \mathsf{Mflow} \cdot \left(\mathsf{h}_3 - \mathsf{h}_2 \right) = 12.599 \, \mathsf{MW}$$

$$\mathsf{Mflow}_{\mathsf{fuel}} \coloneqq \frac{\mathsf{Q}_{\mathsf{input}}}{\mathsf{H}_{\mathsf{fuel}} \cdot \eta_{\mathsf{combustion}}} = 0.291 \, \frac{\mathsf{kg}}{\mathsf{s}}$$

Specific fuel consumption

$$SFC := \frac{Mflow_{fuel}}{Thrust_{total}} = 3.09 \times 10^{-5} \frac{s}{m}$$

Gas temperature as it leaves the jet pipe into the atmosphere

$$T_{6} := T_{5} \cdot \left(\frac{p_{a}}{p_{5}}\right)^{\frac{\gamma_{expand} - 1}{\gamma_{expand}}} = 692.913 \text{ K}$$

Plot Cycle on Temperature-Entropy Plot

Define data to plot

$$Temps := \begin{bmatrix} T_{0'} & T_{1''} & T_{2''} & T_{3''} & T_{4''} & T_{5''} & T_6 \end{bmatrix}$$
 Pressures := $\begin{bmatrix} p_{a'} & p_{1'} & p_{2'} & p_{3'} & p_{4'} & p_{5'} & p_a \end{bmatrix}$

NDEntropies := [seq(Property(entropy, air, P = Pressures[i], temperature = Temps[i]), i = 1..nops(Temps))] p1 := plots:-pointplot(NDEntropies, Temps, connect = true, color = red)

 $\begin{array}{l} p2 := TemperatureEntropyChart(air, \textit{T} = 200 \text{ K}..1200 \text{ K},\\ entropy = 4000 \text{ J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}..5500 \text{ J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}, \text{ isobars = } 0.025 \text{ bar}..20.0 \text{ bar} \end{array}$

